

Applied Science

Advanced GCE

Unit **G628**: Sampling, Testing and Processing

Mark Scheme for June 2013

OCR (Oxford Cambridge and RSA) is a leading UK awarding body, providing a wide range of qualifications to meet the needs of candidates of all ages and abilities. OCR qualifications include AS/A Levels, Diplomas, GCSEs, Cambridge Nationals, Cambridge Technicals, Functional Skills, Key Skills, Entry Level qualifications, NVQs and vocational qualifications in areas such as IT, business, languages, teaching/training, administration and secretarial skills.

It is also responsible for developing new specifications to meet national requirements and the needs of students and teachers. OCR is a not-for-profit organisation; any surplus made is invested back into the establishment to help towards the development of qualifications and support, which keep pace with the changing needs of today's society.

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.




All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

OCR will not enter into any discussion or correspondence in connection with this mark scheme.

© OCR 20

Annotations

Annotation	Meaning
	correct response
	incorrect response
BOD	benefit of the doubt
NBOD	benefit of the doubt not given
ECF	error carried forward
	information omitted
I	ignore
R	reject

Annotation	Meaning
/	alternative and acceptable answers for the same marking point
(1)	separates marking points
not	answers which are not worthy of credit
reject	answers which are not worthy of credit
ignore	statements which are irrelevant
allow	answers that can be accepted
()	words which are not essential to gain credit
—	underlined words must be present in answer to score a mark
ecf	error carried forward
AW	alternative wording
ora	or reverse argument

Question		Answer	Marks	Guidance											
1	(a)	(i)	To protect themselves/danger from falling objects/to conform with health & safety regulations ✓	1	allow prevent injuries										
		(ii)	Does not allow water/liquid through it ✓	1	allow waterproof reject reference to absorption										
		(iii)	Eye protection/gloves/reinforced boots ✓	1											
		(iv)	10g – 2kg inclusive ✓ Needed for repeats/range of tests ✓	1 1	ignore unqualified use ignore reference to representative sampling										
		(v)	any TWO from Location ✓ Health and safety 'label'/hazards ✓ Name of student ✓ Storage information ✓	2	ignore mass/size/appearance/temperature/weather conditions ignore dangerous										
		(vi)	To avoid contamination/loss of sample ✓	1	ignore oxidation/ references to deterioration										
	(b)	(i)	Varied in colour ✓	1	ora										
		(ii)	Brushing/use of solvent which, removes impurities/does not dissolve caliche/use of fan /use of cloth ✓	1	reject water /use of oven ignore unqualified hot air										
		(iii)	<u>Deliquescent</u> ✓	1											
		(iv)	1) 1.2(0) ✓ 2) 0.8(0) ✓	1 1											
(v)		colourless (containing yellow particles) ✓	1	ignore clear											
(vi)		<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">1.17 ✓</td> <td style="width: 33%;">1.18 ✓</td> <td style="width: 33%;">1.2 ✓</td> </tr> <tr> <td>22.5 ± 0.5 ✓</td> <td>23.5 ± 0.5 ✓</td> <td>25.5 ± 0.5 ✓</td> </tr> <tr> <td>$\frac{364.0 \times 22.5}{100} =$</td> <td>$\frac{364.0 \times 23.5}{100} =$</td> <td>$\frac{364.0 \times 25.5}{100} =$</td> </tr> <tr> <td>81.9(0) ✓✓</td> <td>85.5(4) ✓✓</td> <td>92.8(2) ✓✓</td> </tr> </table>	1.17 ✓	1.18 ✓	1.2 ✓	22.5 ± 0.5 ✓	23.5 ± 0.5 ✓	25.5 ± 0.5 ✓	$\frac{364.0 \times 22.5}{100} =$	$\frac{364.0 \times 23.5}{100} =$	$\frac{364.0 \times 25.5}{100} =$	81.9(0) ✓✓	85.5(4) ✓✓	92.8(2) ✓✓	4
1.17 ✓	1.18 ✓	1.2 ✓													
22.5 ± 0.5 ✓	23.5 ± 0.5 ✓	25.5 ± 0.5 ✓													
$\frac{364.0 \times 22.5}{100} =$	$\frac{364.0 \times 23.5}{100} =$	$\frac{364.0 \times 25.5}{100} =$													
81.9(0) ✓✓	85.5(4) ✓✓	92.8(2) ✓✓													

Question		Answer	Marks	Guidance
	(c)	<p>Level 0 (0 marks) Candidate includes fewer than two correct valid points</p> <p>Level 1 (1–2 marks) Candidate gives a description that shows some knowledge of experimental technique but not necessarily in a logical order. For 1 mark candidates should include two valid points. For 2 marks candidates should include at least three valid points.</p> <p>Level 2 (3–4 marks) Candidate gives a description of a workable experiment in a logical and ordered way but in less detail. For 3 marks candidates should include at least four valid points. For 4 marks candidates should include at least five valid points.</p> <p>Level 3 (5–6 marks) Candidate gives a full description of a workable experiment in a logical and well ordered way. For 5 marks candidates should include at least six valid points. For 6 marks candidates should include at seven valid points.</p>	6	<p>valid points to include:</p> <ul style="list-style-type: none"> • Risk assessment/health & safety • Reference to filtering (to remove insoluble particles) • Place solution in a suitable named container (ignore test tube) • Suitable size mentioned ie 100 – 600 cm³ • Reference to partially filling container • Heat /method of heating (to boiling) (reject heat to 300 - 350°C) • Evaporate/reduce (to a small volume) • Leave to cool • Filter (to obtain crystals) • Leave to dry (in the air)/use of oven below 350°C.
	(d) (i)	Risk assessment/PPE/equipment must be clean ✓	1	
	(ii)	Fume cupboard/well ventilated area, as (sulfur dioxide) is a toxic/harmful/irritant gas ✓	1	Both the 'area' in the laboratory and a valid reason must be given for the mark ignore dangerous
	(iii)	0.25(4) ✓ 2.04/2.00✓	2	
	(iv)	Cleaning/washing ✓	1	allow sterilising

Question		Answer	Marks	Guidance
	(e)	1000 x 1000 =, 1 x 10 ⁶ /1 000 000/million ✓	1	
	(f) (i)	30 ✓	1	
	(ii)	70 ✓	1	allow ecf
	(g)	To show the end point of a titration ✓	1	allow 'to show finish/completion' allow neutralisation
	(h) (i)	Plots correct ✓ Suitable straight line ✓	1 1	
	(ii)	correct extrapolation and correct reading ✓	1	allow 0.060 to 0.10 inclusive (plus or minus 1 square)
		Total	36	

Question			Answer	Marks	Guidance
2	(a)	(i)	Pick, flush/sample and test it/check, for insects/disease ✓	1	ignore reference to state of plant allow healthy
		(ii)	Take samples from different, plants/areas ✓	1	
		(iii)	A new flush grows/more growth ✓	1	
		(iv)	They prevent, decay/decomposition/rot/fungi/bacteria ✓	1	ignore reference to chemical reactions
	(b)	(i)	any THREE from Ease of use ✓ Effectiveness/reliability/efficiency/frequency(of insecticide) ✓ Concentration/quantity needed ✓ Health and safety considerations/toxicity ✓ Danger to animals/humans/beneficial insects ✓ Persistence/contact or systemic type/effect on tea taste ✓ Environmental impact ✓ Availability ✓	3	ignore effect on plants/tea
		(ii)	When the insects are feeding/active/attack/around ✓	1	ignore reference to numbers of insects
		(iii)	any ONE from where to spray ✓ when to spray ✓ quantity/concentration of spray ✓ effectiveness of spray on insects ✓ to record concentration ✓ to monitor use/make comparisons ✓	1	ignore which to spray ignore effect on plants
		(iv)	any ONE from greater frequency ✓ more (concentrated) insecticide ✓ change insecticide ✓ change time ✓ method of application ✓	1	allow appropriate reference to faulty equipment
		(v)	Is it 1% of each or 1% in total?/ratio of oil to soap not specified /meaning of concentration/amounts not clear ✓	1	allow is concentration mass/volume or mass/mass etc.

Question		Answer	Marks	Guidance
	(c) (i)	Textbooks/ scientific books/internet/suppliers (catalogues)/ label/ask expert ✓	1	
	(ii)	(amount) too small ✓	1	
	(iii)	any ONE from Time to take effect ✓ Time to degrade/to be removed ✓ No longer harmful for consumption ✓	1	
	(iv)	any ONE from (Safe) disposal ✓ (Safe) storage ✓	1	ignore dilution ignore poured away

Question		Answer	Marks	Guidance
	(d) (i)	300 000 ✓	1	
	(ii)	75 000 ✓	1	ecf on (d) (i) (25% of answer to d(i))
	(e) (i)	any TWO from Choose suitable filter ✓ Zero/reset instrument/use of water ✓ Use of standard solutions/plot calibration curves ✓	2	
	(ii)	<p>Level 0 (0 marks) Candidate includes fewer than two correct valid points</p> <p>Level 1 (1–2 marks) Candidate shows a basic understanding of the procedure. There is minimal structure in the presentation. For 1 mark candidates should include two valid points. For 2 marks candidates should include at least three valid points.</p> <p>Level 2 (3–4 marks) Candidate shows an understanding of the procedure. There is some structure to the presentation. For 3 marks candidates should include at least four valid points. For 4 marks candidates should include at least five valid points.</p> <p>Level 3 (5–6 marks) Candidate shows a high level of understanding and gives a full description of the correct procedure in a logical and well-structured way. Scientific terminology has been used correctly and appropriately. For 5 marks candidates should include at least six valid points. For 6 marks candidates should include at least seven valid points.</p>	6	<p>valid points to include :</p> <ul style="list-style-type: none"> • Add tea bag to the (hot/boiling) water • Use of suitable named container (ignore cuvette) • Reference to size up to 500 cm³ • Maintain constant temperature (use of water bath) • Use of clock • Stir the mixture • Remove samples at suitable time intervals • Allow samples to cool • Place (sample) in cuvette • Place (cuvette) into colorimeter and take reading • Record/compare results (ignore references to graphs)

Question		Answer	Marks	Guidance
	(iii)	Graph goes through origin ✓ Upward curve ✓ Curve becomes flatter/gradient reduced ✓	1 1 1	allow 1 square error curve to be drawn for 3 marks
	(f)	any ONE from (Carbon dioxide) is safe(r)/less harmful ✓ Dichloromethane is toxic ✓ (Carbon dioxide) recovery is easier ✓ (Carbon dioxide) more accessible ✓	1	ignore quicker/easier/reference to taste accept ORA ignore dangerous
	(g)	biodegradable/(easily)decomposes/uses unbleached material /made into compost ✓	1	allow other realistic responses ignore unqualified recycling ignore synthetic
	(h)	100-93/7% 3.5 ✓✓ ecf	2	
		Total	32	

Question			Answer	Marks	Guidance
3	(a)	(i)	a bigger surface area (for extraction)/ expose the inside of the bud/easier/quicker extraction of oil ✓	1	reject references to a chemical reaction
		(ii)	Gases/vapours into liquids/condensation/condenses ✓	1	allow steam
		(iii)	(stops/relieves) pressure build up ✓ Water is forced up the tube (to release the pressure) ✓	2	
		(iv)	any ONE from needs recharging/stopping starting/cleaning ✓ boiler needs refilling ✓ time consuming ✓	1	ignore cost /more people/small amounts
		(v)	Any TWO from Remove stopper and open tap ✓ Run off dichloromethane layer until interface just reaches the tap/lower layer removed ✓ Close tap ✓	2	
		(vi)	Use an evaporating basin/beaker in a fume cupboard /distillation ✓ Use of water bath/controlled electrical heating ✓	2	
		(vii)	$6 \times 0.66 = 4$ ✓ Spot drawn at 4 cm above start line ✓ ecf on calculation	2	allow 3.9 - 4-1
		(viii)	Repeat with a different solvent/adsorption medium ✓	1	
		(ix)	Mr (of the compound is 204) ✓	1	allow molecular/molar mass
		(x)	Infrared/IR (absorption spectroscopy) ✓	1	
	(b)	(i)	Repeat it/reject/treat as an anomaly/outlier ✓	1	
		(ii)	<u>82.0</u> ✓	1	

Question		Answer	Marks	Guidance
	(c) (i)	87 ✓	1	allow values from 86 to 90 inclusive
	(ii)	It is a straight line graph/direct/linear relationship ✓	1	allow increase at a steady rate
	(d) (i)	Any THREE from Volume/quantity of hexane ✓ Volume/quantity of ethanol ✓ Time of stirring ✓ Type of rose petal ✓ Type/method of filtration ✓ Temperature used ✓ Method of evaporation/heat ✓ Where to evaporate/fume cupboard ✓	3	ignore reference to safety precautions ignore solution 2 marks if reference to quantity of solvents
	(ii)	Recovery/reuse (of the solvent) ✓	1	ignore references to small amounts/lower temperature/scaling ignore references to rose petals
		Total	22	

OCR (Oxford Cambridge and RSA Examinations)
1 Hills Road
Cambridge
CB1 2EU

OCR Customer Contact Centre

Education and Learning

Telephone: 01223 553998

Facsimile: 01223 552627

Email: general.qualifications@ocr.org.uk

www.ocr.org.uk

For staff training purposes and as part of our quality assurance programme your call may be recorded or monitored

Oxford Cambridge and RSA Examinations
is a Company Limited by Guarantee
Registered in England
Registered Office; 1 Hills Road, Cambridge, CB1 2EU
Registered Company Number: 3484466
OCR is an exempt Charity

OCR (Oxford Cambridge and RSA Examinations)
Head office
Telephone: 01223 552552
Facsimile: 01223 552553

© OCR 2013

